



# BASE UNITS

## THE SECOND

The second, symbol  $s$ , is the SI unit of time. It is defined by taking the fixed numerical value of the caesium frequency  $\Delta\nu_{\text{Cs}}$ , the unperturbed ground-state hyperfine transition frequency of the caesium-133 atom, to be 9 192 631 770 when expressed in the unit Hz, which is equal to  $s^{-1}$ .

## THE KELVIN

The kelvin, symbol  $K$ , is the SI unit of thermodynamic temperature. It is defined by taking the fixed numerical value of the Boltzmann constant  $k$  to be  $1.380\,649 \times 10^{-23}$  when expressed in the unit  $J\,K^{-1}$ , which is equal to  $kg\,m^2\,s^{-2}\,K^{-1}$ , where the kilogram, metre and second are defined in terms of  $h$ ,  $c$  and  $\Delta\nu_{\text{Cs}}$ .

## THE METRE

The metre, symbol  $m$ , is the SI unit of length. It is defined by taking the fixed numerical value of the speed of light in vacuum  $c$  to be 299 792 458 when expressed in the unit  $m\,s^{-1}$ , where the second is defined in terms of the caesium frequency  $\Delta\nu_{\text{Cs}}$ .

## THE MOLE

The mole, symbol  $mol$ , is the SI unit of amount of substance. One mole contains exactly  $6.022\,140\,76 \times 10^{23}$  elementary entities. This number is the fixed numerical value of the Avogadro constant,  $N_A$ , when expressed in the unit  $mol^{-1}$  and is called the Avogadro number.

The amount of substance, symbol  $n$ , of a system is a measure of the number of specified elementary entities. An elementary entity may be an atom, a molecule, an ion, an electron, any other particle or specified group of particles.

## THE KILOGRAM

The kilogram, symbol  $kg$ , is the SI unit of mass. It is defined by taking the fixed numerical value of the Planck constant  $h$  to be  $6.626\,070\,15 \times 10^{-34}$  when expressed in the unit  $J\,s$ , which is equal to  $kg\,m^2\,s^{-1}$ , where the metre and the second are defined in terms of  $c$  and  $\Delta\nu_{\text{Cs}}$ .

## THE CANDELA

The candela, symbol  $cd$ , is the SI unit of luminous intensity in a given direction. It is defined by taking the fixed numerical value of the luminous efficacy of monochromatic radiation of frequency  $540 \times 10^{12}$  Hz,  $K_{\text{cd}}$ , to be 683 when expressed in the unit  $lm\,W^{-1}$ , which is equal to  $cd\,sr\,W^{-1}$ , or  $cd\,sr\,kg^{-1}\,m^{-2}\,s^3$ , where the kilogram, metre and second are defined in terms of  $h$ ,  $c$  and  $\Delta\nu_{\text{Cs}}$ .

## THE AMPERE

The ampere, symbol  $A$ , is the SI unit of electric current. It is defined by taking the fixed numerical value of the elementary charge  $e$  to be  $1.602\,176\,634 \times 10^{-19}$  when expressed in the unit  $C$ , which is equal to  $A\,s$ , where the second is defined in terms of  $\Delta\nu_{\text{Cs}}$ .