All data taken at Pacific Northwest National Laboratory (PNNL)

Operators: Jerome C. Birnbaum, Tyler O. Danby, Timothy J. Johnson, Molly Rose K. Kelly-Gorham, Rodica Lindenmaier, Tanya L. Myers

SAMPLE CONDITIONS & PHYSICAL PROPERTIES

Chemical name	Sodium chloride
Chemical formula	NaCl
Synonyms	Salt; Halite; Table salt; Rock salt
CAS number	7647-14-5
Location of field sample	n/a
History of sample	n/a
Molecular Weight	58.44 g/mol
Melting Point	801 °C
Boiling Point	1413 °C
Density (25° C)	2.17 g/cm ³
Hardness, Mohs scale	n/a
Crystallography:	
Cell dimension	a = 5.64 Å
Crystal system	cubic
H-M symbol (point gr)	m3m
Space group	225
H-M symbol (space gr)	Fm3m
Crystal habit	Cubes
Color	White
Diaphaneity	Sub-translucent to opaque
Particle size	$314\pm170\;\mu m$
Particle size assessment	Optical microscopy
Supplier	Sigma-Aldrich
Stated purity	≥99.5%
Date packed	18 March 2016 Weight: 3.314 grams
Synthesis method	n/a
Synthesis reference	n/a
Texture	Wet cubic crystals, hygroscopic
Physical state	Crystalline solid
Surface roughness	n/a
Elemental composition	n/a
Isotopic composition	n/a
Moisture content	n/a
Temperature of sample	25 ± 2 °C
Substrate	n/a

INSTRUMENT PARAMETERS

Tensor 37 FT-IR manufactured by Bruker Optics

External diffuse reflectance accessory	A 562-G integrating sphere
Sphere diameter	75 mm
Angle to normal incidence	14.8°
Sphere opening diameter	19 mm (entrance port)
Spectral range	7,500 to 600 cm ^{-1} saved; 7500 to 600 cm ^{-1} reported
Beamsplitter	Ge on KBr
Detector (dia. Det. Port in sphere)	2×2 mm, 60° field of view MCT (550; 0.9); 1 cm
Apodization function	Blackman-Harris 3-term
Aperture	6 mm
Coadded scans	2048
Scanner speed	40 kHz
Switch gain on	512 points
Low pass filter	Open
Scan technique	double-sided, forward-backward
Non-linear correction	Off
High and low folding limit	$15802.38-0.00 \text{ cm}^{-1}$
Phase resolution	32.00
Phase correction mode	Mertz
Zerofilling	$4 \times$
Wavenumber accuracy	$\pm 0.4 \text{ cm}^{-1}$
Spectral resolution	4 cm^{-1}
Accuracy verification	10/28/2015
Wavelength vetted on:	ICL polystyrene standard #0009-7394-0025A, thin film
Reflectance:	$\pm 2\%$ using SRS reflectance standards 50-010-DH27B-4878

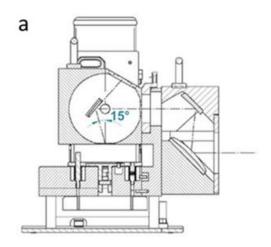




Figure 1: The Bruker 562-G integrating sphere (a) and Tensor 37 (b)

Photographs of sample Sodium Chloride

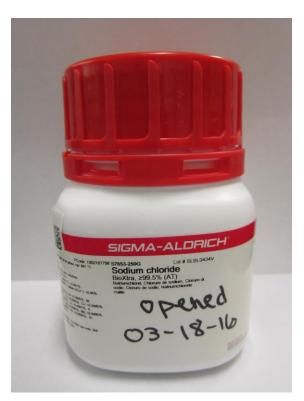


Figure 2: Sodium Chloride in Sigma-Aldrich container.



Figure 3: Sodium Chloride loaded in IR sample cup.

PARTICLE SIZE PREPARATION AND CHARACTERIZATION

Optical microscopy —

A Keyence VHX-1000 digital microscope with 16-bit resolution is used to provide photomicrographs of the various samples and particle sizes. Software included with the microscope differentiates the brightness and colors in the image and extracts the bright objects to produce a binary image. The software assumes all adjacent bright points are part of the same object then calculates the area for each of these objects. The area (A) is used to calculate the mean particle diameter (d) by assuming the particles are spherical and using the relationship $d=(4*A/\pi)1/2$. Although the assumption of spherical particles is clearly not always valid, this procedure provides a reasonable estimate of the mean particle size.

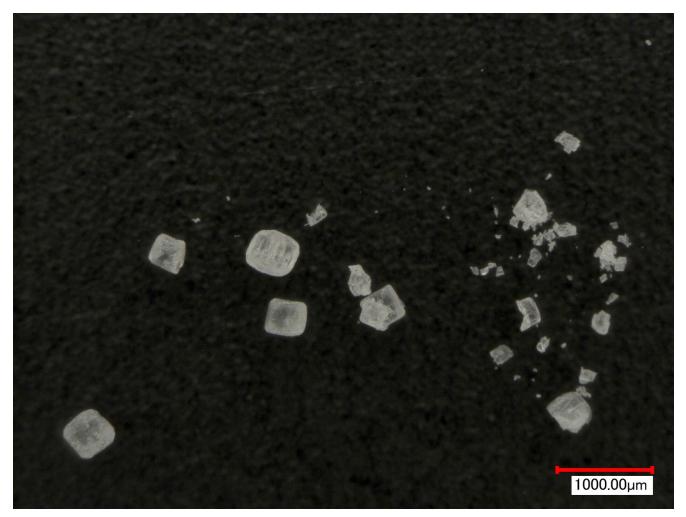


Figure 4: Photomicrograph of Sodium Chloride.

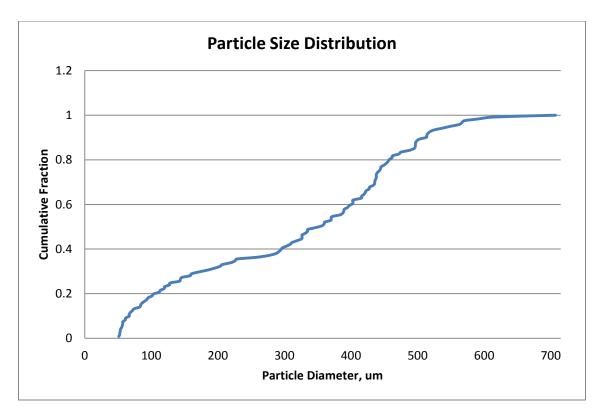


Figure 5: Particle size distribution of Sodium Chloride.