All data taken at Pacific Northwest National Laboratory (PNNL)

Operators: Jerome C. Birnbaum, Tyler O. Danby, Timothy J. Johnson, Molly Rose K. Kelly-Gorham, Rodica Lindenmaier, Tanya L. Myers

SAMPLE CONDITIONS & PHYSICAL PROPERTIES

Chemical name Ammonium phosphate monobasic

Chemical formula H₆NO₄P

Synonyms mono-Ammonium phosphate; Ammonium dihydrogenphosphate; prim-

Ammonium phosphate

CAS number 7722-76-1

Location of field sample n/a History of sample n/a

Molecular Weight 115.03 g/mol

Melting Point 190 °C Boiling Point n/a

Density (20° C) 1.81 g/cm³

Hardness, Mohs scale n/a

Crystallography:

Cell dimension $a = \mathring{A} b = \mathring{A} c = \mathring{A}$

Crystal system

H-M symbol (point gr)

Space group

H-M symbol (space gr)

Crystal habit

Color White/clear

Diaphaneity Transparent to sub-translucent

Particle size $414 \pm 269 \, \mu m$ Particle size assessment Optical microscopy Supplier Sigma-Aldrich Stated purity $\geq 99.99\%$

Date packed 06 July 2016 Weight: 2.416 grams

Synthesis method n/a Synthesis reference n/a

Texture Hard, shiny crystals

Physical stateSolidSurface roughnessn/aElemental compositionn/aIsotopic compositionn/aMoisture contentn/aTemperature of sample 25 ± 2 °CSubstraten/a

INSTRUMENT PARAMETERS

Tensor 37 FT-IR manufactured by Bruker Optics

External diffuse reflectance accessory A 562-G integrating sphere

Sphere diameter 75 mm Angle to normal incidence 14.8°

Sphere opening diameter 19 mm (entrance port)

Spectral range 7,500 to 600 cm⁻¹ saved; 7500 to 600 cm⁻¹ reported

Beamsplitter Ge on KBr

Detector (dia. Det. Port in sphere) 2×2 mm, 60° field of view MCT (550; 0.9); 1 cm

Apodization function Blackman-Harris 3-term

Aperture 6 mm

Coadded scans 2048

Scanner speed 40 kHz

Switch gain on 512 points

Low pass filter Open

Scan technique double-sided, forward-backward

Non-linear correction On

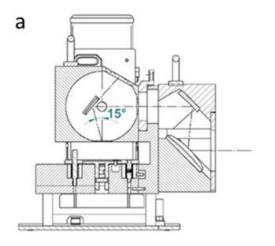
High and low folding limit 15800.54-0.00 cm⁻¹

Phase resolution 32.00Phase correction mode MertzZerofilling $4 \times$

Wavenumber accuracy $\pm 0.4 \text{ cm}^{-1}$ Spectral resolution 4 cm^{-1}

Accuracy verification 10/28/2015

Wavelength vetted on: ICL polystyrene standard #0009-7394-0025A, thin film Reflectance: ±2% using SRS reflectance standards 50-010-DH27B-4878



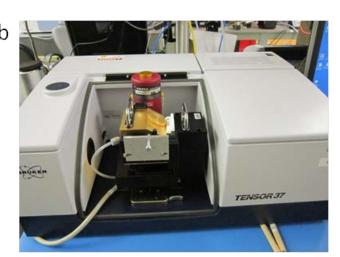


Figure 1: The Bruker 562-G integrating sphere (a) and Tensor 37 (b)

Photographs of sample Ammonium phosphate monobasic



Figure 2: Ammonium phosphate monobasic in Sigma-Aldrich container.

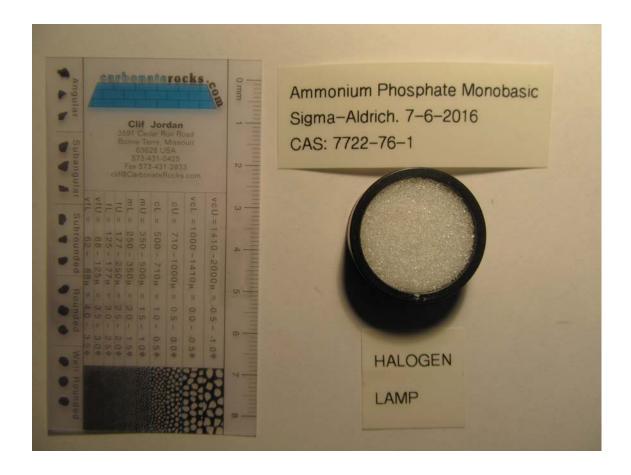


Figure 3: Ammonium phosphate monobasic loaded in IR sample cup.

PARTICLE SIZE PREPARATION AND CHARACTERIZATION

Optical microscopy —

A Keyence VHX-1000 digital microscope with 16-bit resolution is used to provide photomicrographs of the various samples and particle sizes. Software included with the microscope differentiates the brightness and colors in the image and extracts the bright objects to produce a binary image. The software assumes all adjacent bright points are part of the same object then calculates the area for each of these objects. The area (A) is used to calculate the mean particle diameter (d) by assuming the particles are spherical and using the relationship $d=(4*A/\pi)1/2$. Although the assumption of spherical particles is clearly not always valid, this procedure provides a reasonable estimate of the mean particle size.

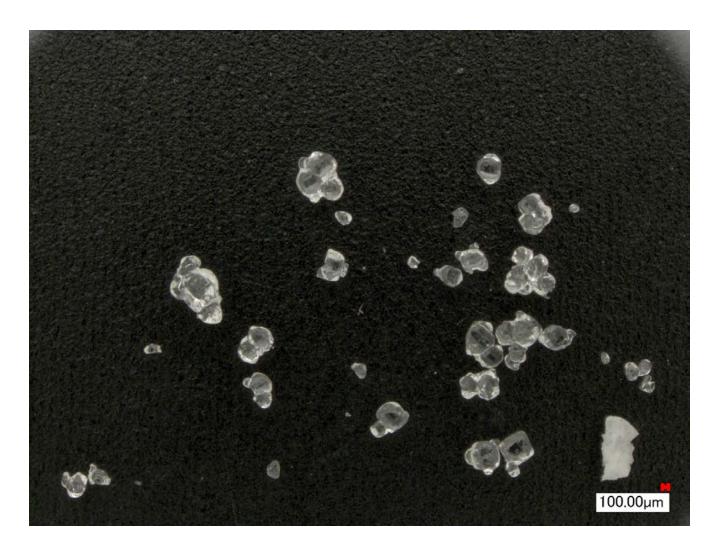


Figure 4: Photomicrograph of Ammonium phosphate monobasic.

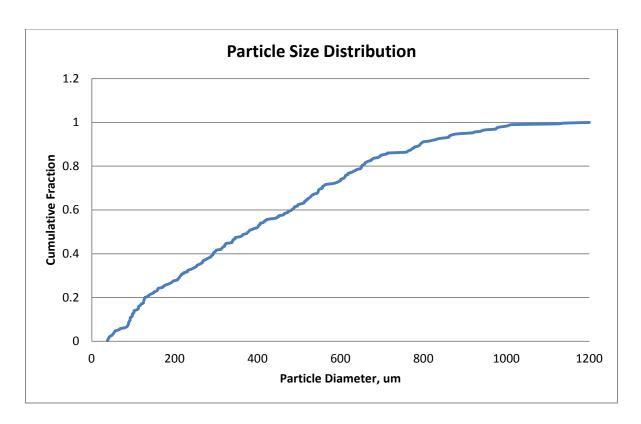


Figure 5: Particle size distribution of Ammonium phosphate monobasic.