All data taken at Pacific Northwest National Laboratory (PNNL)

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SAMPLE CONDITIONS & PHYSICAL PROPERTIES

Chemical name	2-Naphthol
Chemical formula	$C_{10}H_7OH$
Synonyms	β-Naphthol; 2-Hydroxynaphthalene
CAS number	135-19-3
Location of field sample	n/a
History of sample	n/a
Molecular Weight	144.17 g/mol
Melting Point	120-122 °C
Boiling Point	285-286 °C
Density (20 ° C)	1.28 g/cm^3
Hardness, Mohs scale	n/a
Crystallography:	
Cell dimension	a = A b = A c = A
Crystal system	
H-M symbol (point gr))
Space group	
H-M symbol (space gr)
Crystal habit	
A 1	
Color	Light salmon
Diaphaneity	Light salmon Opaque
	Opaque Bimodal distribution: $69 \pm 31 \ \mu m$ for fine particles (<500 μm); 3559 ± 1316
Diaphaneity Particle size	Opaque Bimodal distribution: $69 \pm 31 \ \mu m$ for fine particles (<500 $\ \mu m$); $3559 \pm 1316 \ \mu m$ for large particles (>500 $\ \mu m$)
Diaphaneity Particle size Particle size assessment	Opaque Bimodal distribution: $69 \pm 31 \mu\text{m}$ for fine particles ($<500 \mu\text{m}$); $3559 \pm 1316 \mu\text{m}$ for large particles ($>500 \mu\text{m}$) Optical microscopy
Diaphaneity Particle size Particle size assessment Supplier	Opaque Bimodal distribution: $69 \pm 31 \ \mu m$ for fine particles (<500 $\ \mu m$); $3559 \pm 1316 \ \mu m$ for large particles (>500 $\ \mu m$) Optical microscopy Aldrich
Diaphaneity Particle size Particle size assessment Supplier Stated purity	Opaque Bimodal distribution: $69 \pm 31 \mu\text{m}$ for fine particles ($<500 \mu\text{m}$); $3559 \pm 1316 \mu\text{m}$ for large particles ($>500 \mu\text{m}$) Optical microscopy
Diaphaneity Particle size Particle size assessment Supplier Stated purity Date packed	OpaqueBimodal distribution: 69 ± 31 μm for fine particles (<500 μm); 3559 ± 1316
Diaphaneity Particle size Particle size assessment Supplier Stated purity Date packed Synthesis method	Opaque Bimodal distribution: $69 \pm 31 \ \mu m$ for fine particles ($<500 \ \mu m$); $3559 \pm 1316 \ \mu m$ for large particles ($>500 \ \mu m$) Optical microscopy Aldrich 99% 01 September 2016 Weight: 1.564 grams n/a
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INSTRUMENT PARAMETERS

IR Cube FT-IR manufactured by Bruker Optics

External diffuse reflectance accessory	
External diffuse reflectance accessory	A 562-G integrating sphere
Sphere diameter	75 mm
Angle to normal incidence	14.8°
Sphere opening diameter	19 mm (entrance port)
Spectral range	7,500 to 600 cm^{-1} saved; 7500 to 600 cm^{-1} reported
Beamsplitter	Ge on KBr
Detector (dia. Det. Port in sphere)	2×2 mm, 60° field of view MCT (550; 0.9); 1 cm
Apodization function	Blackman-Harris 3-term
Aperture	Open
Coadded scans	1024
Scanner speed	40 kHz
Switch gain on	512 points
Low pass filter	Open
Scan technique	double-sided, forward-backward
Non-linear correction	On
High and low folding limit	$15800.54-0.00 \text{ cm}^{-1}$
Phase resolution	32.00
Phase correction mode	Mertz
Zerofilling	4×
Wavenumber accuracy	$\pm 0.4 \text{ cm}^{-1}$
Spectral resolution	4 cm^{-1}
Accuracy verification	11/17/2015
Wavelength vetted on:	ICL polystyrene standard #0009-7394-0025A, thin film
Reflectance:	±2% using SRS reflectance standards 50-010-DH27B-4878

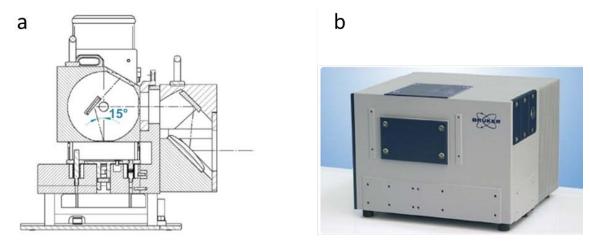


Figure 1: The Bruker 562-G integrating sphere (a) and IR Cube (b)

Photographs of sample 2-Naphthol



Figure 2: 2-Naphthol in Aldrich container.

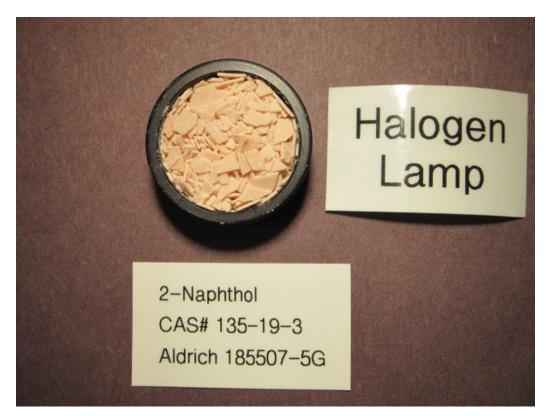


Figure 3: 2-Naphthol loaded in IR sample cup.

PARTICLE SIZE PREPARATION AND CHARACTERIZATION

Optical microscopy -

A Keyence VHX-1000 digital microscope with 16-bit resolution is used to provide photomicrographs of the various samples and particle sizes. Software included with the microscope differentiates the brightness and colors in the image and extracts the bright objects to produce a binary image. The software assumes all adjacent bright points are part of the same object then calculates the area for each of these objects. The area (A) is used to calculate the mean particle diameter (d) by assuming the particles are spherical and using the relationship $d=(4*A/\pi)1/2$. Although the assumption of spherical particles is clearly not always valid, this procedure provides a reasonable estimate of the mean particle size.



Figure 4: Photomicrograph of 2-Naphthol.

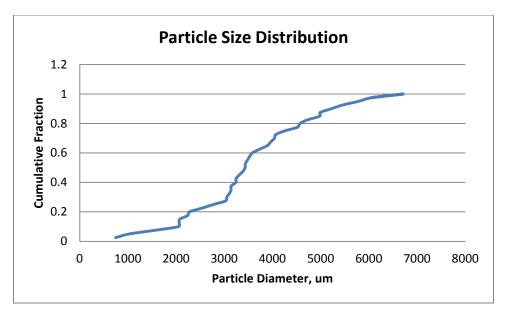


Figure 5: Particle size distribution of 2-Naphthol for large particles (> $500 \mu m$).

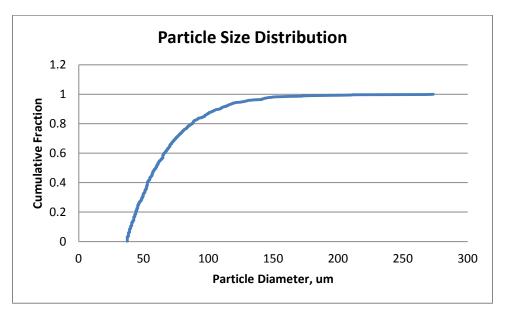


Figure 6: Particle size distribution of 2-Naphthol for fine particles ($< 500 \,\mu$ m).